

Development of CO₂ Reduction Catalysts to Mitigate Global Warming and Energy Crisis

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Abstract

The increasing anthropogenic emission of CO₂ has caused many global environmental issues i.e. global warming, ocean acidification and species extinction.

Conversion of CO₂ to useful chemicals including fuels utilizing clean and sustainable solar light becomes a promising approach that can not only tackle CO₂ emission caused global issues but also provide a clean and sustainable energy source simultaneously.

In this project, a series of novel multinuclear metal complexes consists of earth abundance metals and bridging polypyridyl ligands will be designed and synthesized as potential CO₂ reduction photocatalysts.

The catalytic performance of the new metal complexes will also be studied using (i) Gas Chromatography with Thermal Conductivity Detector (GC-TCD) for CO and H₂ generation, (ii) ion chromatography for formate ion.

Introduction

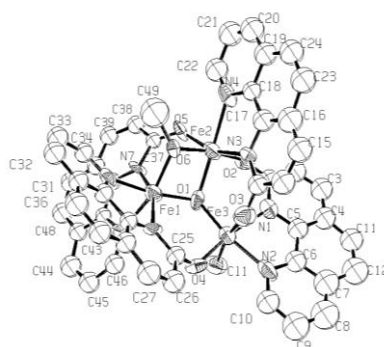
The acceleration of artificial photosynthesis research in recent years have revealed the really crucial and challenging issues in photoconversion of CO₂ into valuable chemicals and fuels. Some primary challenging issues include the extremely complicated photocatalytic reaction mechanisms and pathways, low efficiency and selectivity of different products, complex activation and adsorption of CO₂ molecules and richness of excited state dynamics and the semiconductor surface chemistry.

Objective

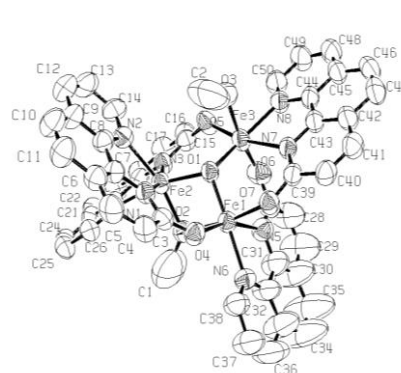
- To design, synthesize and characterized a series of earth abundant metal complexes with polypyridyl ligands
- To study the CO₂ reduction catalytic properties of the novel complexes.
- To design, develop and study the photocatalytic systems systemically and rationally.

Results

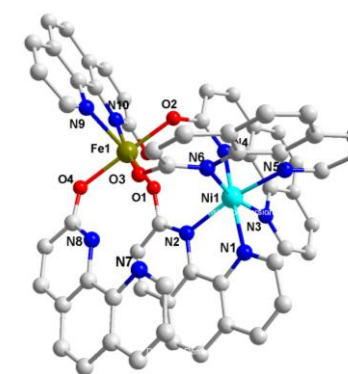
Many new multinuclear polypyridyl transitional metal complexes have been successfully synthesized and characterized by IR spectroscopy, mass spectrometry and x-ray crystallography. The ORTEP diagrams are shown below.



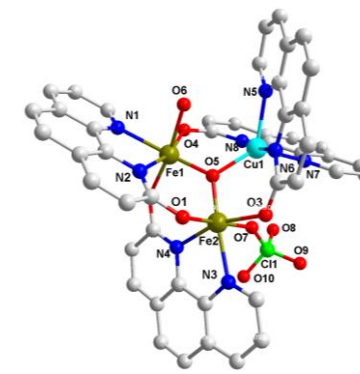
$[(\mu_3\text{-O})(\mu\text{-OMe})\text{Fe}_3(\mu\text{-Phen-O})_4(\text{OMe})]^+$



$[(\mu_3\text{-O})(\mu\text{-OMe})\text{Fe}_3(\mu\text{-Phen-O})_4\text{Cl}]^+$



$[\text{FeNi}(\text{PhenO})_3(\text{PhenOH})_2]^{2+}$



$[\text{CuFe}_2\text{O}(\text{PhenO})_4(\text{ClO}_4)(\text{H}_2\text{O})]^+$

Acknowledgements

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